SULFURIC ACID

Information Fire and Explosion

Fire:

SECTION 2

Not combustible, but substance is a strong oxidizer and its heat of reaction with reducing agents or combustibles may cause ignition. Not combustible, but substance is a strong oxidizer and its heat of reaction with reducing agents or combustibles may cause ignition. Reacts with most metals releasing flammable, potentially explosive hydrogen gas.

Fire Extinguishing Media:

Explosion:

Dry chemical, foam or carbon dioxide. Water spramay be used to keep fire exposed containers cool. Water spray

Special Information:

In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode.

Reactivity Data

Stability:

Stable under ordinary conditions of use and storage.

SECTION 3

Hazardous Decomposition Products:

Toxic fumes of oxides of sulfur. Will react with water or steam to produce toxic and corrosive fumes. Reacts with carbonates to generate carbon dioxide gas, and with cyanides and sulfides to form poisonous hydrogen cyanide and hydrogen sulfide respectively.

Hazardous Polymerization:

Incompatibilities:

Will not occur

Water, bases, organic material, halogens, metal acetylides, oxides and hydrides, strong oxidizing and reducing agents and many other reactive substances

Leak/Spill Disposal Information SECTION 4

Dike and cover leaking or spilled liquid with dirt, vermiculite, kitty-litter or other inert absorbent. Cover spill with sodium bicarbonate or soda ash and mix. Clean-up personnel require protective clothing and respiratory protection from vapors and mists. Neutralized waste may be containerized and disposed in a RCRA approved waste disposal facility. Flush area of spill with dilute soda ash solution and discard to sewer.

Reportable Quantity (RQ)(CWA/CERCLA) : 1000 lbs

Ensure compliance with local, state and federal regulations

Hazardous Ingredients: Not applicable. Formula CAS No.: 7664-93-9 Synonyms: Oil of Vitriol

Effective Date: 08-26-85 PRODUCT IDENTIFICATION:

Molecular Weight: 98.07 Chemical Formula: H2SO4

CAUSES SEVERE BURNS. PRECAUTIONARY MEASURES MAY BE FATAL IF SWALLOWED. HARMFUL IF INHALED.

Do not get in eyes, on skin, or on clothing.

Do not breathe mist.

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Keep container closed.

Use only with adequate ventilation.

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Wash thoroughly after handling.

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This substance is classified as a POISON under the Federal Caustic Poison Act. EMERGENCY/FIRST AID

In all cases call a physician. flush skin or eyes with plenty of water for at In case of contact, immediately flush skin or eyes with plenty of water for at In case of contact. If swallowed, DO NOT INDUCE VOMITING! Give large quantities of least 15 minutes.

Water: Never give anything by mouth to an unconscious person. If inhaled, remove water: Never give anything by mouth to artificial respiration. If breathing is to fresh air. difficult, give oxygen.

SEE SECTION 5.

DOT Hazard Class: Corrosive Material

SECTION 1

Appearance: Colorless, oily liquid

Physical Data

Odor:

Infinite @ 20°C.

Solubility:

Specific Gravity: 1.84

Boiling Point: ca 310°C (590°F)

Melting Point: ca -14°C (6°F)

Vapor Density (Air=1):< 0.3 @ 25°C (77°F)

Vapor Pressure (mm Hg):1 @ 146°C (250°F).

Evaporation Rate: No information found.